WITHANOLIDES OF ACNISTUS BREVIFLORUS

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Key Word Index—Acnistus breviflorus; Solanaceae; withanolides; steroidal lactones.

Abstract—The chemical examination of *Acnistus breviflorus* afforded nine withanolides, four of which are new and were established as 2,3,24,25-tetrahydro-27-desoxywithaferin A (4β -hydroxy- 5β , 6β -epoxy-1-oxo-22R-withanolide), 2,3-dihydro-27-desoxywithaferin A (4β -hydroxy- 5β , 6β -epoxy-22R-witha-24-enolide), 5,6-desoxywithaferin A (4β ,27-dihydroxy-1-oxo-22R-witha- 2β ,24-dienolide). The five known compounds were: withaferin A; 2,3-dihydrowithaferin A; 24,25-dihydro- 2β -desoxywithaferin A and withaferin A- 2β -chlorohydrin.

INTRODUCTION

As part of our continued interest in the withanolides of the plants belonging to the Solanaceae [1], we undertook a systematic chemical examination of Acnistus breviflorus with the objective of isolating their components and laying the groundwork for understanding their biosynthesis. Nine withanolides were obtained during liquid chromatography of the chloroform extract, four of which are new compounds. The compounds occurring in the largest amounts were withaferin A (1a) [2] and 2,3dihydrowithaferin A (2a) [2, 3]. Other known compounds present in minor quantities were 24,25-dihydro-27desoxywithaferin A (3a) [4], 27-desoxywithaferin A (4) [4] and 6α -chloro- 5β -hydroxywithaferin A (5) [5, 6]. The four new compounds were established by spectral analysis as: 2,3,24,25-tetrahydro-27-desoxywithaferin A (6a); 2,3dihydro-27-desoxywithaferin A (7a); 5,6-desoxywithaferin A (8a) and 2,3-dihydro-5,6-desoxywithaferin A (9a).

RESULTS

Among the members of the Solanaceae family, Withania sommifera is so far outstanding in the elaboration of a variety of withanolides [1]. This species occurs as different chemotypes and was found to be a good model for the study of chemogenetics [7]. Along these lines, we undertook the chemical analysis of other members of this family. The genus Dunalia is sometimes synonymous with Acnistus [8] and it is possible that a member of this genus could produce such a model. Acnistus australis from Argentina [9], A. arborescens from Brazil [10] and, more recently, A. ramiflorum [11] have already been investigated. Hence, we chose A. breviflorus, originating from Argentina and grown in our experimental plots, for our study. The structure elucidation of the new compounds is presented in the sequel.

Compound **6a** ($C_{28}H_{42}O_5$) was established as 2,3,24,25-tetrahydro-27-desoxywithaferin A. In the UV, this compound did not show the usual absorptions for the unsaturated ketone of ring A and δ -lactone of the sidechain; however, the IR showed their carbonyl groups, the frequencies suggesting them to be saturated. The ¹³C NMR spectrum showed no sp^2 carbons in addition to

these two carbonyls, thus double bonds are absent. The protons of the two secondary methyl groups of the reduced lactone (Table 1) are in accord with similar ones in the known compound 3a [4, 12] and, by correlation, have also similar conformations, 27-Me is β -eq and 28-Me is α -eq, as drawn. The 4β -hydroxyl gave an acetate (6b), a compound also obtained by catalytic hydrogenation of the known 3b. The mass spectrum of 6a showed a fragment at m/z 331 [M - 127] which was attributed to the loss of the saturated side-chain by cleavage of the C-17,C-20 bond [4].

Compound 7a (C₂₈H₄₀O₅) was established as 2,3dihydro-27-desoxywithaferin A. Four sp^2 carbons were shown by the ¹³C NMR spectrum, while the ¹H NMR spectrum indicated the absence of the usual enone system of ring A but revealed a close similarity with the structure of 6a. The acetate 7b was compared with a synthetic sample obtained from withaferin A diacetate (1b) by careful hydrogenation (over 5% Pd-CaCO₃), whereby the Δ^2 was saturated and the allylic 27-OAc underwent hydrogenolysis [13]. Compound 7a was hydrogenated (over 10% Pd-C) and the product (10) had the two lactonic secondary methyl groups in a cis-orientation $(J_{24,25} = 14.4 \,\mathrm{Hz})$ in contrast to their trans stereochemistry observed ($J_{24,25} = 10 \,\text{Hz}$, [12]) in the natural compounds. In fact, the various possible conformers of the saturated lactones in withanolides have been discussed [14] and by correlation, in 10, the 27-Me is β -eq and 28-Me is β -ax.

Compound 8a ($C_{28}H_{38}O_5$) was established as 5,6-desoxywithaferin A. The ¹³C NMR spectrum showed eight sp^2 carbons. Thus, a trisubstituted double bond (corresponding to one singlet and one doublet in the SFORD spectra) could be present in addition to the usual unsaturated systems of ring A and the side-chain (UV and IR spectra). This double bond could be placed at C-5,C-6, as deduced from an analysis of the ¹H NMR spectrum (Table 1) which compared well with that of withanolide U(11) [15] for protons in the A/B rings system. For confirmation, 11 was reacted with MnO₂. The double allylic 4β -hydroxyl was oxidized to the ketone 12 thus resulting in a diene-dione in rings A/B, $\lambda_{\rm max}$ 225 and 278 nm (ϵ 21 000 and 2200). A similar oxidation of 8a

Table 1. ¹H NMR signals of relevant protons in withanolides

					Methyl groups			groups			
Compound											
-	2-Н 3-Н	4-H	6-H	22-Н	18	19	21	27	28	Other signals	
6a		3.52	3.14	4.32	0.67	1.32	0.94	1.12	1.22		
		(t, 3.1)	$(br \ s)$	(dt, 11.6, 3.2)			(d, 6.7)	(d, 6.7)	(d, 6.7)		
6b		4.34	3.18	4.36	0.67	1.33	0.95	1.14	1.23	2.06 (4-OAc)	
		(m)	$(br\ s)$	(dt, 11.6, 3.2)			(d, 6.7)	(d, 6.7)	(d, 6.7)		
7a		3.52	3.20	4.35	0.67	1.32	0.99	1.88	1.93		
		(m)	(br s)	(dt, 13.2, 3.4)			(d, 6.7)				
		4.57	3.15	4.35	0.67	1.28	0.98	1.88	1.93	2.06 (4-OAc)	
7b		(t, 3.2)	$(br\ s)$	(dt, 13.2, 3.4)			(d, 6.7)			•	
)		3.51	3.14	4.33	0.66	1.32	0.95	1.14	0.93		
10		(m)	$(br\ s)$	(m)	****		(d, 7)	(d, 7)	(d, 7)		
	5.96 6.77	4.63	5.93	4.45	0.76	1.57	1.04		2.05	4.38 (m, 27-CH ₂ OH)	
8a	(d, 9.8)(dd, 10			(dt, 13.1, 3.5)	0.70		(d, 6.8)				
,	6.02 6.71	5.88	6.11	4.43	0.75	1.39	1.03	_	2.06	2.09 (27-OAc and 4-OAc)	
	(d, 10) (dd, 10		$(br \ d \ 4.5)$	(dt, 13.1, 3.5)	00	2.07	(d, 6.4)			4.90 (d, 2.1, 27-CH ₂ OAc)	
9a	(a, 10) (au, 10	3.64	5.75	4.43	0.73	1.52	1.02		2.05	4.36 (d, 4.4, 27-CH ₂ OH)	
		(m)	(dd, 4.4, 0.6)	(dt, 13.2, 2.4)	0.75	1.02	(d, 6.8)			(,,	
,		5.43	5.93	4.43	0.72	1.42	1.02	_	2.06	2.09 (27-OAc), 2.08 (4-OAc)	
'			$(br \ d, 4.5)$	(dt, 13.2, 2.4)	0.72	1.72	(d, 6.6)		2.00	4.90 (d, 2.1, 27-CH ₂ OAc)	
	6.91 (dd, 5.2,	2.4) (m)	6.71	4.22	1.08	1.41	1.30	1.89	1.96	, o (a, 2.1, 2. O. 5.2 o. 10)	
l	0.91 (aa, 5.2,	2.4) —	(dd, 10.4, 4.3)	(dd, 13.1, 3.7)	1.00	1.71	1.50	1.07	1.70		
,	606 (3.06)	_	3.42	4.35	0.72	1.01	1.38	1.88	1.94		
3	6.86 (d, 0.6)		$(br \ d \ 2.1)$	(dt, 13.2, 3.5)	0.72	(d, 6.4)	1.50	1.00	1.74		
	670 (44.20.2	6.2)	(<i>br a 2.</i> 1) 6.84	(at, 13.2, 3.3) 4.35	0.76	1.39	1.04		2.05	4.38 (d, 4.3, 27-CH ₂ OH)	
ļ	6.70 (dd, 20.3	, 6.2) —			0.70	1.37	(d, 6.4)		2.03	7.50 (u, 7.5, 27-CH2OH)	
	(70 (11 22 2	(2)	(dd, 5.8, 2.4)	(dt, 13.2, 3.4)	0.78	1.39	1.33		2.55	10.26 (s, 27-СНО)	
;	6.70 (dd, 20.3	, 6.2) —	6.83	4.35	0.78	1.39			4.33	10.20 (3, 27-0110)	
			(dd, 5.6, 2.6)	(dt, 13.2, 3.4)	0.67	1 1 4	(d, 7)	1.00	1.04	3.09 (4.OAs)	
i		4.34		4.36	0.67	1.14	0.96	1.88	1.94	2.08 (4-OAc)	
		(m)		(dt, 13.1, 3.3)			(d, 6.3)				

Chemical shifts are in δ units; coupling constants (in Hz) are in parentheses.

$$\mathbf{1a} \quad \mathbf{R} = \mathbf{H}, \, \mathbf{R}' = \mathbf{OH}$$

1b
$$R = Ac$$
, $R' = OAc$

2a 2,3-dihydro,
$$R = H$$
, $R' = OH$

2b 2.3-dihydro,
$$R = Ac$$
, $R' = OAc$

$$4 \quad R = R' = H$$

$$7a$$
 2,3-dihydro, $R = R' = H$

7b 2.3-dihydro,
$$R = Ac$$
, $R' = H$

$$3a R = H$$

$$3b R = Ac$$

6a 2,3-dihydro,
$$R = H$$

6b 2,3-dihydro,
$$R = Ac$$

$$R''$$
 CH_2R'
 R''

8a
$$R = R' = OH, R'' = H$$

8b
$$R = R' = OAc$$
, $R'' = H$

9a 2.3-dihydro,
$$R = R' = OH$$
, $R'' = H$

9b 2,3-dihydro,
$$R = R' = OAc$$
, $R'' = H$

11
$$R = R'' = OH, R' = H$$

12
$$\Delta^5$$
, R = OH, R' = Me

13 5,6β-epoxy,
$$R = H$$
, $R' = Me$

14
$$\Delta^{5}$$
, R = H, R' = CH₂OH
15 Δ^{5} , R = H, R' = CHO

15
$$\Lambda^5$$
, R = H, R' = CHO

produced compounds 14 and 15, in which an identical system is present. However, the latter also contained an aldehyde, resulting from the oxidation of the allylic 27-CH₂OH ($\nu_{\rm max}$ 2878 and 1690 cm⁻¹ for the -CHO group). This aldehyde also induces a deshielding of the 28-Me by 0.6 ppm.

When the diacetate **8b** was hydrogenated over Pd, 3 mol of H_2 were absorbed resulting in the product **16** with saturation of the Δ^2 bond, hydrogenolysis of 27-CH₂OAc and Δ^5 reduction in the given sequence.

Compound 9a $(C_{28}H_{40}O_5)$ was established as 2,3-dihydro-6,7-desoxywithaferin A. This compound was lacking the usual enone system of ring A, as shown by the ¹H NMR spectrum. The ¹³C NMR spectrum indicated six sp^2 carbons assigned to the two carbonyl and two double bonds, one being at Δ^5 . The 4β -OH is allylic to the latter, which is exo-trisubstituted and as such could not be oxidized with MnO₂, but was easily acetylated. The diacetate 9b absorbed 2 mol of H₂, producing a product (16) found identical in all respects with that obtained from 8b

DISCUSSION

It was found that all the compounds isolated from Acnistus breviflorus (1-9) have a hydroxyl group at C-4 but none at C-20, C-17 and C-14, and that the substitution pattern in rings A/B, unlike the situation in Withania somnifera [1], lacks variety. A relatively new feature seems to be dominant in this plant, namely the reduction of the double bonds. For example, the yield of 2,3-dihydrowith a ferin A (2a) (50 g), compared to that of with a ferin A (1a) (90 g), from 12 kg of the plant material indicates the predominance of this reaction in this plant. The occurrence of a saturated lactone (in compounds 3a and **6a)** is a characteristic which is not widespread among the withanolides and has so far been encountered only in a South African type of W. somnifera [4]. Interestingly, during a hybridization of this plant with chemotype II (Israel), this reduction became a dominant character [12, 16]. These results indicate that an appropriate genetic combination may lead to such a dominant character as encountered in Acnistus breviflorus. The occurrence of a reduced lactone, together with a nonreduced ring A, indicates a biosynthetic process which cannot be duplicated in the laboratory, the sequence of reduction being inverse.

The isolation of 5,6-desoxywithaferin A (8a) which seems to be a direct precursor of the naturally abundant withaferin A (1a) in this plant, lends support to the biogenetic scheme presented earlier [1].

EXPERIMENTAL

Mps were measured on a Fischer–Johns apparatus and are uncorr. Optical rotations were determined in CHCl₃. IR spectra were recorded in KBr pellets; UV spectra were recorded for EtOH solns; ¹H NMR spectra were determined on a Bruker WH270 instrument and ¹³C NMR spectra on a Brucker WH90, operating at 22.63 MHz in CDCl₃ with TMS as internal standard. For LC, a 500-g Si gel G 60 (E. Merck) column was eluted with hexane–EtOAc mixtures and 11. fractions were collected. Mass spectra were determined under the direction of Dr. Z. Zaretskii, and microanalyses carried out by Mr. R. Heller of our Institute.

Isolation of the withanolides from Acnistus brevifiorus Griseb. The extraction procedure, carried out on 12kg leaves, was described earlier [6]. The withanolides were isolated from the

various fractions by CC of the CHCl₃ extract as presented below. The fractions were worked up separately to yield the indicated compounds:

Frac-	Eluant (hexane-EtOAc)	Compound	Total yield (g)
17–25	1:1	5	0.02
29-35	1:1	Mixture A $(4 + 6a)$	0.50
33-48	1:1	7a	0.75
45-60	2:3	Mixture B $(8a + 9a)$	0.42
53-69	2:3	5	0.45
67-95	2:3	1	90.0
88-120	2:3	2a	50.0

 4β -Hydroxy- 5β , 6β -epoxy-1-oxo-22R-withanolide (6a). Mixture A contained, in equal proportions, 4 and 6a, which had identical R_f 0.65 value on TLC plates (EtOAc). A small quantity of pure 4 (100 mg) was obtained after several rapid crystallizations from hot MeOH. The mother liquors were evapd to dryness (400 mg), dissolved in EtOAc-CHCl₃ (2:1, 200 ml) and shaken for 6 hr. with freshly prepared MnO₂ (300 mg). The residue from the filtrate, upon separation by prep. TLC gave two products: unreacted compound 6a (210 mg) and compound 13 (85 mg), the oxidation product of 4. Compound 6a mp 209-210° from EtOAc; $[\alpha]_D + 17.4^{\circ} (c \, 0.2)$; IR $v_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$: 3420, 1730, 1705, 1440, 1370, 1190, 1020, and 920; 13 C NMR: δ 211.6 (C-1), 72.9 (C-4), 66.7 (C-5), 58.8 (C-6), 11.6 (C-18), 15.4 (C-19), 14.5 (C-21), 78.3 (C-22), 176.6 (C-26), 12.8 (C-27) and 21.5 (C-28); MS m/z(rel. int.): 458 $[M]^+$ (5.7), 440 $[M-H_2O]^+$ (7.5), 442 $[M - 2 H_2O]^+$ (3.9), 331 $[M - 127]^+$ (5.1), 313 $[M - 127]^+$ $[H_2O]^+$ (4.2) and 127 (63.2). (Found: C, 73.3; H, 9.35; $C_{28}H_{42}O_5$ requires C, 73.42 and H, 9.24%.) The acetate 6b, obtained by pyridine-Ac₂O; crystals, mp 215-217° from EtOAc; $[\alpha]_D$ $+25.2^{\circ}$ (c 0.1). (Found: C, 71.5; H, 8.98; C₃₀H₄₄O₆ requires C, 72.06 and H, 8.87%.) Compound **3b** on hydrogenation (5%) Pd-CaCO₃) in EtOH also gave 6b, compared by IR and NMR spectra.

 5 β,6β-Epoxy-1,4-dioxo-22R-witha-2,24-dienolide (13). Mp 252-253° from EtOAc; UV $\lambda_{\text{max}}^{\text{EiOH}}$ nm: 287.5 and 223 (ε 1900 and 20 100); IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1698, 1685, 1670, 1440, 1260, 1110 and 900; MS m/z (rel. int.): 452 [M]⁺ (0.9), 284 (22), 270 (5.5), 256 (16), 185 (7.5) and 125 (13.5). (Found: C, 74.19; H, 8.42; $C_{28}H_{36}O_5$ requires C, 74.4 and H, 8.03 %.)

 4β -Hydroxy-5β,6β-epoxy-1-oxo-22R-witha-24-enolide (7a). Mp 236–237° from EtOAc; $[\alpha]_D$ – 3.0° (c 0.12); UV λ_{\max}^{EIOH} nm: 224 (ε 7800); IR ν_{\max}^{KBr} cm $^{-1}$: 3400, 1715, 1697, 1380, 1185, 942, and 915; 13 C NMR: δ 66.7 (C-5), 58.5 (C-6), 11.5 (C-18), 15.2 (C-19), 13.4 (C-21), 78.3 (C-22), 149.2 (C-24), 121.9 (C-25), 167.1 (C-26), 12.5 (C-27) and 20.5 (C-28); MS m/z (rel. int.): 456 [M] $^+$ (12), 438 [M - H₂O] $^+$ (18), 420 [M - 2 H₂O] $^+$ (8.5), 331 [M - 125] $^+$ (12), 313 [M - 125 - H₂O] $^+$ (6), 301 (8.5), 288 (51.5), 225 (12.5), 181 (88) and 125 (100). (Found: C, 73.42; H, 8.98; C₂₈H₄₀O₅ requires C, 73.75 and H, 8.35%.) The acetate 7b, obtained by pyridine–Ac₂O treatment; crystals, mp 246–247° from EtOAc; $[\alpha]_D$ – 3.8° (c 0.1); UV λ_{\max}^{EIOH} nm: 224.5 (ε 7200); IR ν_{\max}^{KBr} cm $^{-1}$: 1735, 1720, 1706, 1235, 1120, 1020 and 950. (Found: C, 72.12; H, 8.72; C₃₀H₄₂O₆ requires C, 72.35 and H, 8.5%.)

 4β -Hydroxy-5 β ,6 β -epoxy-1-oxo-22R-24-isowithanolide (10). Compound 7a (50 mg) in EtOH (25 ml) was hydrogenated over 10 % Pd-C (50 mg) for 8 hr and the product (45 mg), mp 227-228° crystallized from EtOAc. (Found: C, 73.27; H, 9.45; C₂₈H₄₂O₅ requires C, 73.42 and H, 9.24%.)

4β,27-Dihydroxy-1-oxo-22R-witha-2,5,24-trienolide Mixture B contained, in equal proportions, 8a and 9a with identical R_f 0.53 value on TLC (EtOAc) which co-crystallized from various solvents. Compound 8a was obtained pure after several rapid crystallizations from hot MeOH (50 mg); mp 218–220°; $[\alpha]_D + 42.5^\circ$ (c 0.17); UV λ_{max}^{EtOH} nm 215: (ϵ 17 000); IR v_{max}^{KBr} cm $^{-1}$: 3400, 1700, 1665, 1393, 1265, 1130, 1010 and 963; 13 C NMR: δ 202.6 (C-1), 131.0 (C-2), 142.9 (C-3), 69.2 (C-4), 129.1 (C-5), 126.1 (C-6), 11.8 (C-18), 22.8 (C-19), 13.3 (C-21), 78.8 (C-22), 153.0 (C-24), 125.7 (C-25), 167.1 (C-26), 57.3 (C-27) and 20.0 (C-28); m/z (rel. int.): 454 [M] (25.4%), 436 [M - H₂O]⁺ (26.2), 283 $[M - 151 - H_2O]^+$ (16), 265 $[M - 151 - 2H_2O]^+$ (13), 239 (13.3), 225 (31.1), 169 (13.2) and 157 (31.7). (Found: C, 74.01; H, 8.48; $C_{28}H_{38}O_5$ requires C, 74.07 and H, 8.44%.) The diacetate 8b, obtained by pyridine-Ac2O treatment, mp 252–253° from EtOAc; $[\alpha]_D + 48.2^\circ (c\ 0.11)$; UV λ_{max}^{EtOH} nm: 214.5 (ε 18 000). (Found: C, 71.26; H, 8.09; C₃₂H₄₂O₇ requires C, 71.43 and H, 7.85%.)

 4β ,27-Dihydroxy-1-oxo-22R-witha-5,24-dienolide (9a). The mother liquors from mixture B were evapd and the residue oxidized with MnO₂ as before. The product separated into three components by prep. TLC with EtOAc as eluant: unreacted compound 9a (190 mg) and the two oxidation products 14 (45 mg) and 15 (15 mg). Compound 9a, mp 212-213° from EtOAc; $[\alpha]_D + 27.4^{\circ} (c\ 0.2)$; UV λ_{max}^{EtOH} nm: 225 (ϵ 8800); IR ν_{max}^{KBr} cm $^{-1}$: 3400, 1718, 1680, 1410, 1200, 1025 and 976; 13 C NMR: δ 214.8 (C-1), 73.5 (C-4), 143.7 (C-5), 128.0 (C-6), 12.0 (C-18), 21.4 (C-19), 13.5 (C-21), 79.1 (C-22), 154.1 (C-24), 125.8 (C-25), 167.5 (C-26), 57.2 (C-27) and 12.0 (C-28); MS m/z (rel. int.): 456 [M]⁺ (20), 438 [M - H₂O]⁺ (15.5), 285 (16), 267 (18), 227 (16.6), 199 (20.7), 185 (17) and 159 (33.6). (Found: C, 73.42; H, 8.98; $C_{28}H_{34}O_5$ requires C, 73.75 and H, 8.84%.) The diacetate, **9b**, obtained by pyridine-Ac₂O treatment; crystals, mp 228-230° from EtOAc; $[\alpha]_D + 35.7^\circ$ (c 0.13); UV λ_{max}^{EtOH} nm: 224 (ϵ 8600). (Found: C, 71.42; H, 8.38; C₃₀H₄₂O₆ requires C, 71.17 and H, 8.21 %.)

27-Hydroxy-1,4-dioxo-22R-witha-2,5,24-trienolide (14). Mp 245–247° from EtOAc; UV $\lambda_{\rm max}^{\rm EIOH}$ nm: 287 and 222 (ϵ 1800 and 22 000); IR $\nu_{\rm max}^{\rm KB}$ cm $^{-1}$: 3400, 1698, 1676, 1668, 1625, 1450, 1260 and 1020. (Found: C, 74.12; H, 8.23, C₂₈H₃₆O₅ requires C, 74.4 and H, 8.03%.)

1,4,27-Trioxo-22R-witha-2,5,24-trienolide (15). Mp 215–216° from EtOAc; UV λ_{max}^{EtOH} nm: 289 and 222 (ϵ 3200 and 25 400); IR ν_{max}^{KBr} cm $^{-1}$: 2878, 1700, 1690, 1678, 1668, 1530, 1450 and 1375. (Found: C, 74.33; H, 7.81, $C_{28}H_{34}O_5$ requires C, 74.73 and H, 7.62%.)

 $14\alpha,20\alpha_{\rm F}$ -Dihydroxy-1,4-dioxo-22R-witha-2,5,24-trienolide (12). Withanolide U (11) (100 mg) was oxidized with MnO₂ (300 mg) as before and the product was recrystallized from EtOAc, mp 274–275°. [α]_D + 35.1° (c 0.11); UV $\lambda_{\rm max}^{\rm EcO}$ nm: 225 and 278 (ϵ 21 000 and 2200); IR $\nu_{\rm max}^{\rm KB}$ cm $^{-1}$:3510,1700, 1685, 1662, 1620, 1600, 1430 and 1120. (Found: C, 71.62; H, 7.82; C₂₈H₃₆O₆ requires C, 71.85 and H, 7.75%.) MS m/z (rel. int.): 468 [M] $^+$ (21), 450 [M - H₂O] $^+$ (6), 325 [M - 125 - H₂O] $^+$ (99.5), 307 [M - 125 - 2 H₂O] $^+$ (16), 187 (44.5), 169 (28), 126 (100), 125 (90).

4-Acetoxy-1-oxo-22R-witha-24-enolide (16). Compounds 8b and 9b (40 mg) in EtOH (30 ml) were hydrogenated in two

separate batches over 5% Pd-CaCO₃ for 10 hr when **8b** absorbed 3 mol and **9b**, 2 mol of H₂. From both, the same product **16**, was obtained (35 mg), mp 229-230° from EtOAc; MS m/z (rel. int.) 486 [M]⁺ (0.1), 444 [M - COCH₂]⁺, (0.5), 426 [M - AcOH]⁺ (52), 408 [M - AcOH - H₂O]⁺ (16.6), 305 [M - 125 - AcOH]⁺ (11), 287 [M - 125 - AcOH - H₂O]⁺ (17), 125 (100). (Found: C, 74.22; H, 9.25. $C_{30}H_{44}O_{5}$ requires C, 74.44; H, 9.16%.)

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NOTE ADDED IN PROOF

Recently it has been reported that jaborosalactone A and acnistoferin have been isolated from a specimen of A. breviftorus collected in Tucumán, Argentina, which is probably a different chemotype. These withanolides have no hydroxyl group at C-4 [17].